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(R,S)-1,4-Bis(benzylsulfinyl)butane

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Key indicators

Single-crystal X-ray study $T=293~{\rm K}$ Mean $\sigma({\rm C-C})=0.007~{\rm \AA}$ R factor = 0.068 wR factor = 0.165 Data-to-parameter ratio = 16.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The molecular structure of the title compound, $C_{18}H_{22}O_2S_2$, exhibits an R,S configuration for the S atoms. The molecule sits on a centre of symmetry located at the midpoint of the central C-C bond. The C and S atoms between the two phenyl rings are essentially coplanar and make a 87.3 (3)° dihedral angle with the phenyl ring planes.

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Comment

Disulfoxides are known to be multifunctional ligands that can coordinate to metal ions *via* either their O or S atoms, depending on electronic and steric factors. The inherent chiral property of the S atom can lead to the formation of diastereomeric *meso* and *rac* isomers. These features make such compounds very interesting ligands for constructing framework complexes. During our investigations of the coordination chemistry of disulfoxides, several ligands of the type $RS(O)(CH_2)_nS(O)R$ have been structurally determined (Li *et al.*, 2002, 2004; Chen *et al.*, 2001), which exist in diastereo-isomeric *rac* or *meso* forms with different melting points.

As shown in Fig. 1, the title compound, (I), has an *R*,*S* configuration based on the S atoms. The molecule has a centre of symmetry at the midpoint of the central C—C bond. A similar situation had been observed in (*R*,*S*)-1,2-bis(methylsulfiny)ethane (Svinning *et al.*, 1976), (*R*,*S*)-1,2-bis(phenylsulfiny)ethane (Cattalini *et al.*, 1979) and (*R*,*S*)-1,2-bis(benzylsulfinyl)ethane (Li *et al.*, 2002). All the C and S atoms between the two phenyl rings are almost coplanar, having an average deviation of 0.0287 (3) Å. The two S—O groups lie on opposite sides of the plane and their pseudotorsion angle (S—O···S—O) is 180°, as required by symmetry. The dihedral angle between this plane and each phenyl ring is 87.3 (3)°. The bond parameters are comparable with the corresponding values for analogous compounds, *viz*.

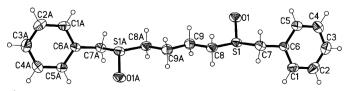


Figure 1 View of the title compound, shown with 50% displacement ellipsoids. Symmetry code: A - x, 1 - y, 2 - z.

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organic papers

(*R*,*S*)-1,2-bis(benzylsulfinyl)ethane (Li *et al.*, 2002) and (*R*,*S*)-1,2-bis(phenylsulfiny)ethane (Cattalini *et al.*, 1979).

Experimental

(R,S)-1,4-Bis(benzylsulfinyl)butane was obtained according to the procedure of Zhang *et al.* (1995) (yield: 40%, m.p. 442–444 K). IR (KBr pellet, cm $^{-1}$): 3062 (*w*), 3031 (*w*), 2958 (*m*), 2920 (*m*), 2860 (*w*), 1603 (*w*), 1495 (*m*), 1454 (*s*), 1415 (*m*), 1285 (*w*), 1159 (*w*), 1087 (*w*), 1072 (*m*), 1023 (*vs*), 915 (*w*), 889 (*w*), 768 (*s*), 697 (*s*), 670 (*w*), 572 (*w*), 497 (*m*), 431 (*w*); ¹H NMR (CDCl₃): δ 1.60–1.93 (4H, *m*), 2.55 (4H, *t*), 3.99 (4H, *q*), 7.26–7.28 (4H, *m*), 7.34–7.41 (6H, *m*). Colourless single crystals were obtained by recrystallization from chloroform/methanol (1:1 v/v).

Crystal data

$C_{18}H_{22}O_2S_2$	$D_x = 1.275 \text{ Mg m}^{-3}$
$M_r = 334.48$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 928
a = 19.382 (8) Å	reflections
b = 5.303 (2) Å	$\theta = 3.2 - 20.0^{\circ}$
c = 8.492 (3) Å	$\mu = 0.31 \text{ mm}^{-1}$
$\beta = 93.126 (6)^{\circ}$	T = 293 (2) K
$V = 871.5$ (6) \mathring{A}^3	Block, colourless
Z = 2	$0.36 \times 0.28 \times 0.24 \text{ mm}$

Data collection

Bruker SMART 1000 CCD area- detector diffractometer φ and ω scans Absorption correction: multi-scan ($SADABS$; Bruker, 1998) $T_{min} = 0.897$, $T_{max} = 0.929$	1611 independent reflections 977 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.054$ $\theta_{\text{max}} = 25.5^{\circ}$ $h = -23 \rightarrow 23$ $k = -6 \rightarrow 2$
$T_{\text{min}} = 0.897$, $T_{\text{max}} = 0.929$ 4306 measured reflections	$k = -6 \to 2$ $l = -10 \to 10$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0713P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.069$	+ 0.483P]
$wR(F^2) = 0.165$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} < 0.001$
1611 reflections	$\Delta \rho_{\text{max}} = 0.55 \text{ e Å}^{-3}$
100 parameters	$\Delta \rho_{\min} = -0.30 \text{ e Å}^{-3}$
H-atom parameters constrained	

Table 1 Selected geometric parameters (Å, °).

S1-O1 S1-C8	1.494 (3) 1.767 (5)	S1-C7	1.806 (4)
O1-S1-C8 O1-S1-C7	105.8 (2) 108.4 (2)	C8-S1-C7	98.0 (2)

H atoms were placed in calculated positions [C-H = 0.93 and 0.97 Å, and $U_{\rm iso}({\rm H})$ = 1.2 $U_{\rm eq}({\rm C})$] and were included in the refinement in the riding-model approximation.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1998); data reduction: *SAINT* and *SHELXTL* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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References

Bruker (1998). SMART (Version 5.051), SAINT (Version 5.01), SADABS (Version 2.03) and SHELXTL (Version 6.1). Bruker AXS Inc., Madison, Wisconsin, USA.

Cattalini, L., Michelon, G., Marangoni, G. & Pelizzi, G. (1979). J. Chem. Soc. Dalton Trans. pp. 96–101.

Chen, W., Du, M., Weng, W., Zhang, R. H. & Bu, X. H. (2001). Acta Cryst. E57, o430–o431.

Li, J. R., Bu, X. H. & Zhang, R. H. (2004). Inorg. Chem. 43, 237-244.

Li, J.-R., Zhang, R.-H. & Bu, X.-H. (2002). Acta Cryst. E58, o911–o912.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Svinning, T., Mo, F. & Bruun, T. (1976). Acta Cryst. B32, 759-764.

Zhang, R. H., Zhan, Y. L. & Chen, J. T. (1995). Synth. React. Inorg. Met. Org. Chem. 25, 283–292.